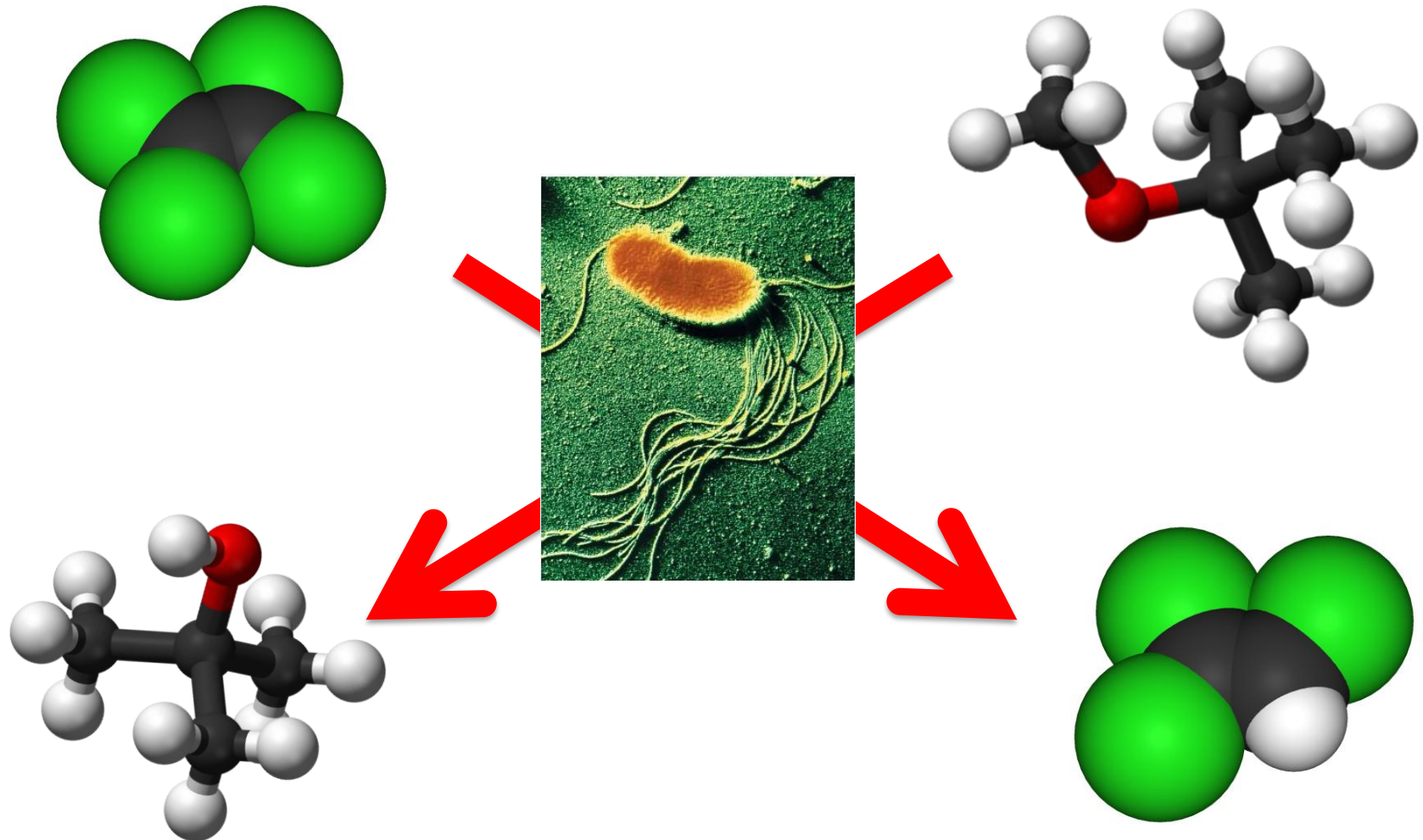


Biodegradation of chlorinated solvents and fuel oxygenates



Outline

- **Chlorinated solvents--PCE and TCE**
 - Properties of pollutants and abiotic processes
 - Aerobic cometabolism and anaerobic metabolism
- **Fuel oxygenates--MTBE and TBA**
 - Properties of pollutants
 - Anaerobic and aerobic metabolism
 - Aerobic cometabolism
- **Fluorinated analogs as EAPS**

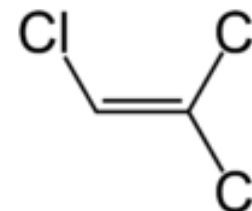
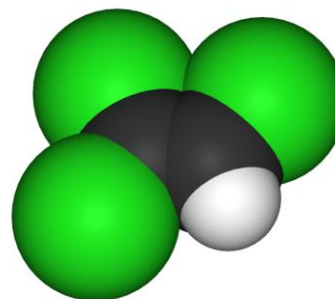
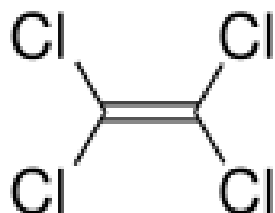
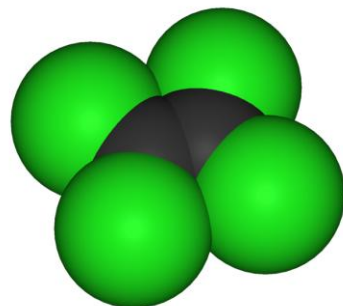
Chlorinated solvents

PCE and TCE



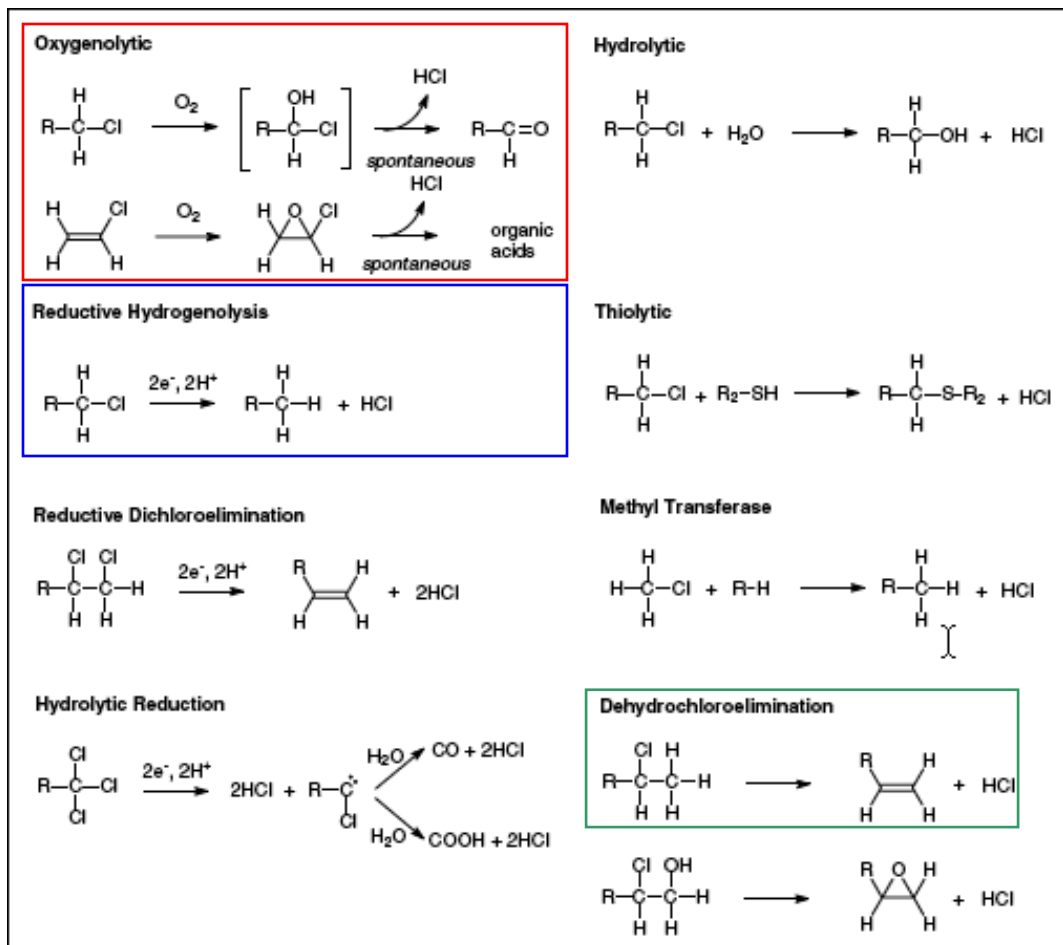
Overview of PCE and TCE

- **PCE: (perchloroethylene, perc, tetrachloroethene)**
 - High volume chemical used as degreasing and dry cleaning solvent
- **TCE: (trichloroethylene, trike, 1,1,2-trichloroethene)**
 - Historically widely used as a solvent.



- Both common ground water pollutants and both are persistent under aerobic conditions
- No organisms know to use either compound as electron donors.
- PCE: undergoes limited abiotic hydrolysis; PCE and TCE undergo abiotic reduction

Dehalogenation mechanisms



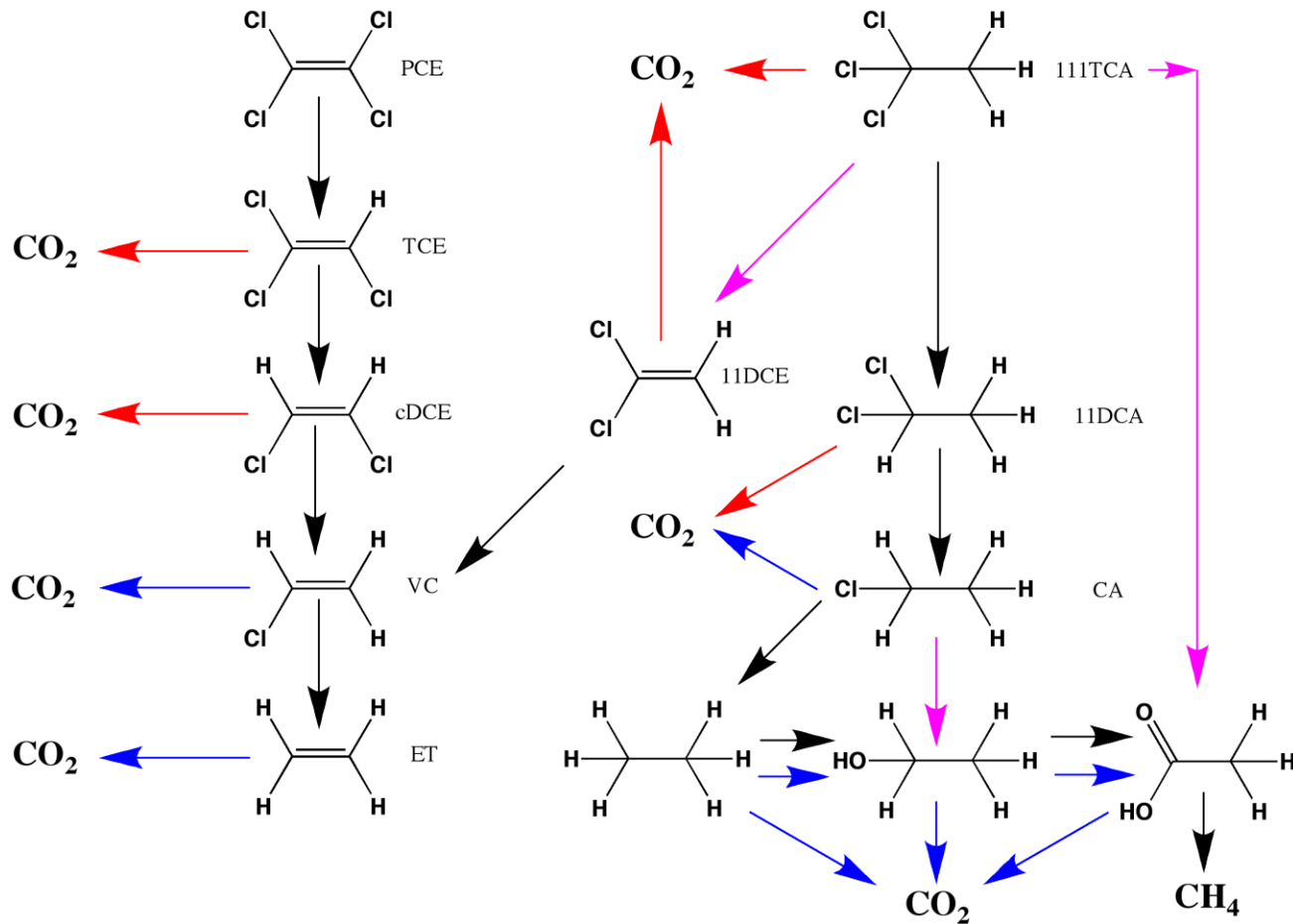
Many and varied. Some biotic, some abiotic, some both!

Oxygenolytic: Biodegradation process that requires O₂ for mono- or dioxygenases.

Reductive hydrogenolysis (*aka reductive dehalogenation*): Most common anaerobic biodegradation mechanism for PCE and TCE. Also occurs abiotically.

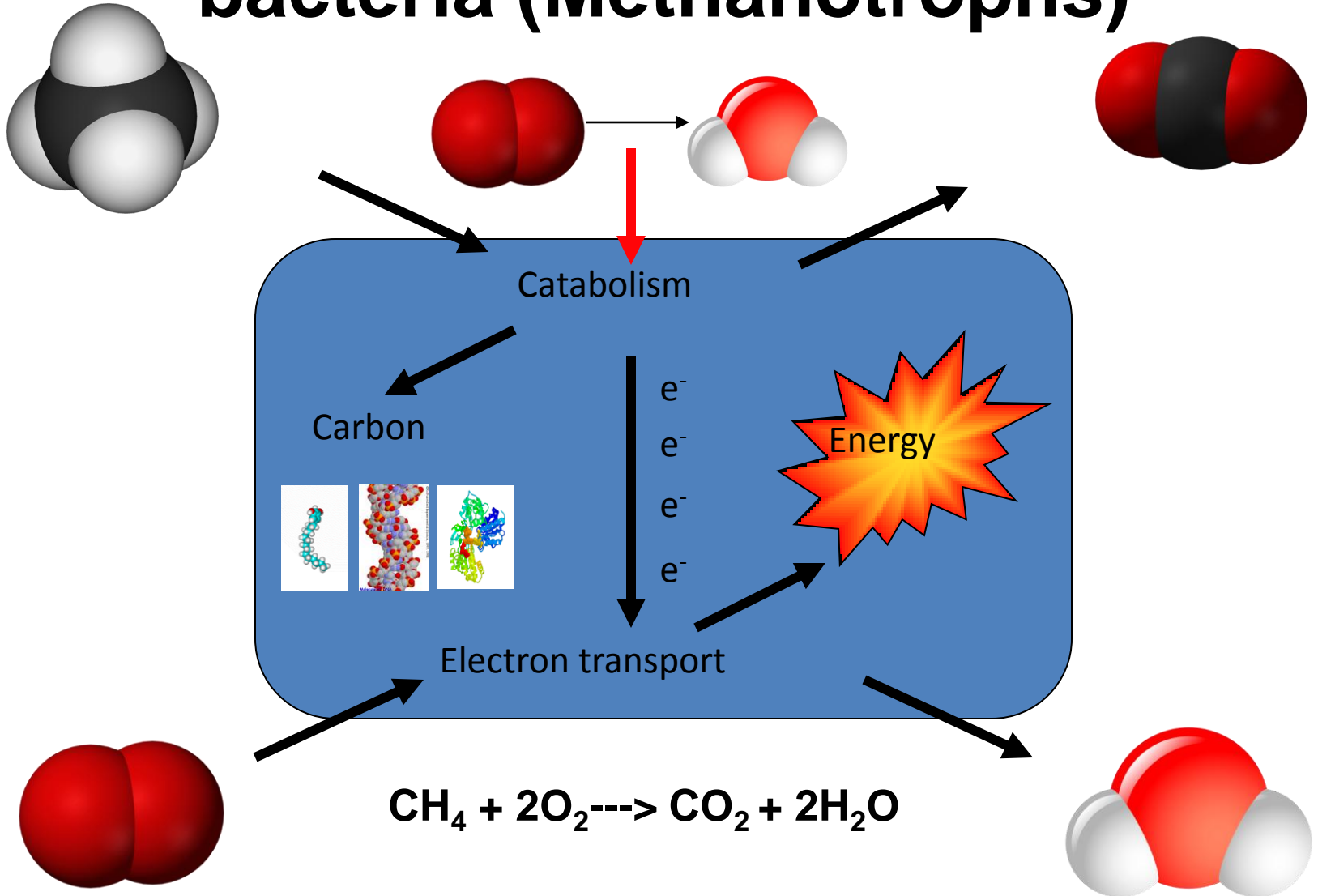
Dehydrochloroelimination: Common abiotic process that converts haloalkanes to haloalkenes

Abiotic and Biotic Transformations

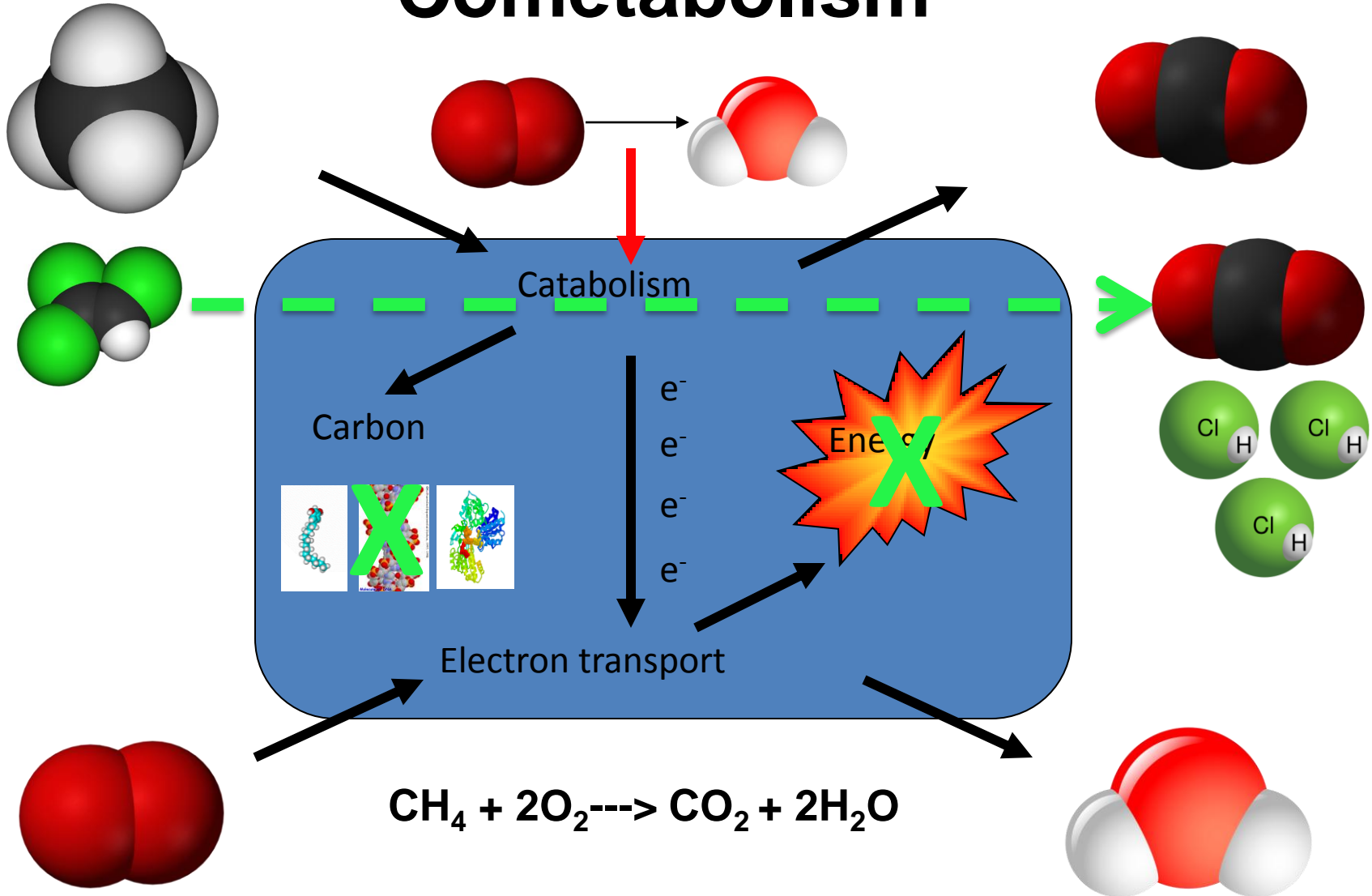


| | | | |
|---------------------------------|-------------------------------|-------------------------------------|------------------------------------|
| ← | ← | ← | ← |
| Aerobic Cometabolism | Aerobic Metabolism | Reductive Dehalogenation | Abiotic Transformations |

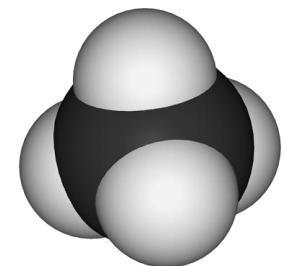
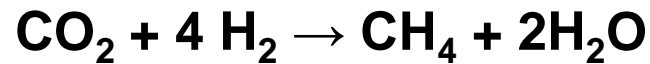
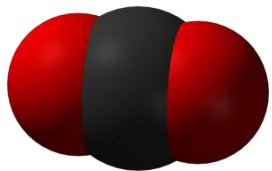
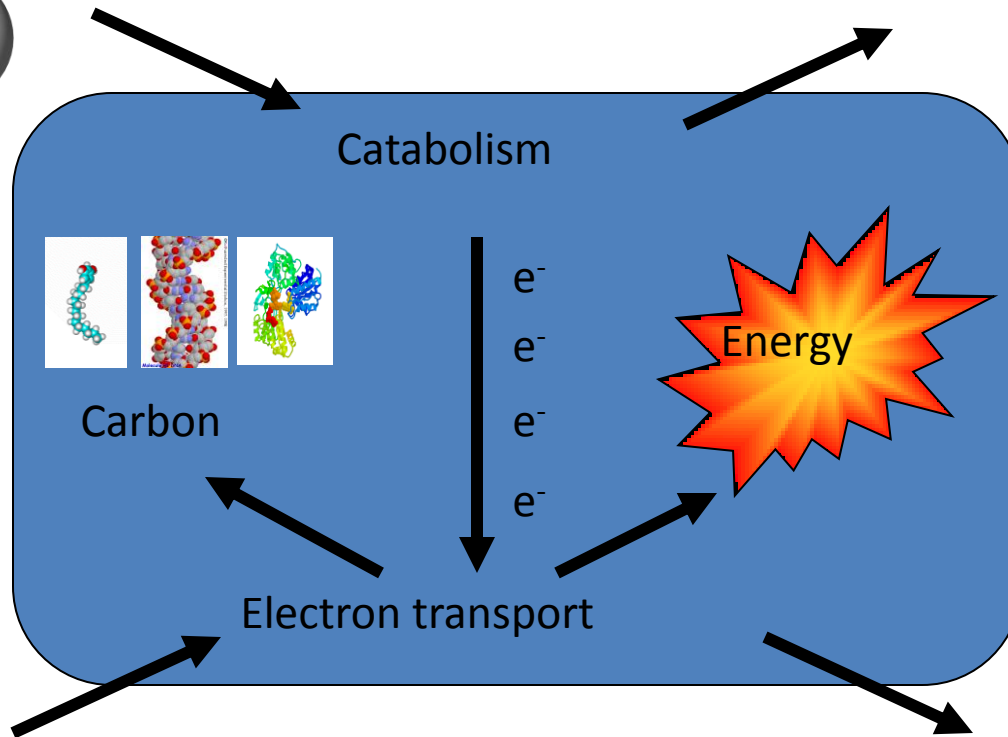
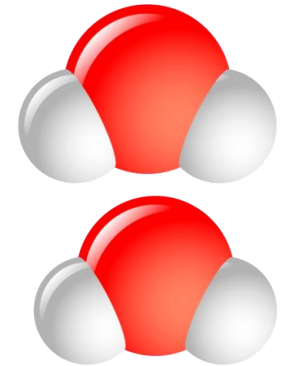
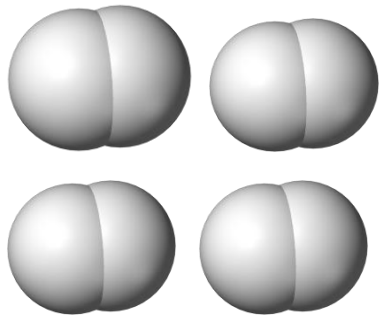
Aerobic methane-oxidizing bacteria (Methanotrophs)



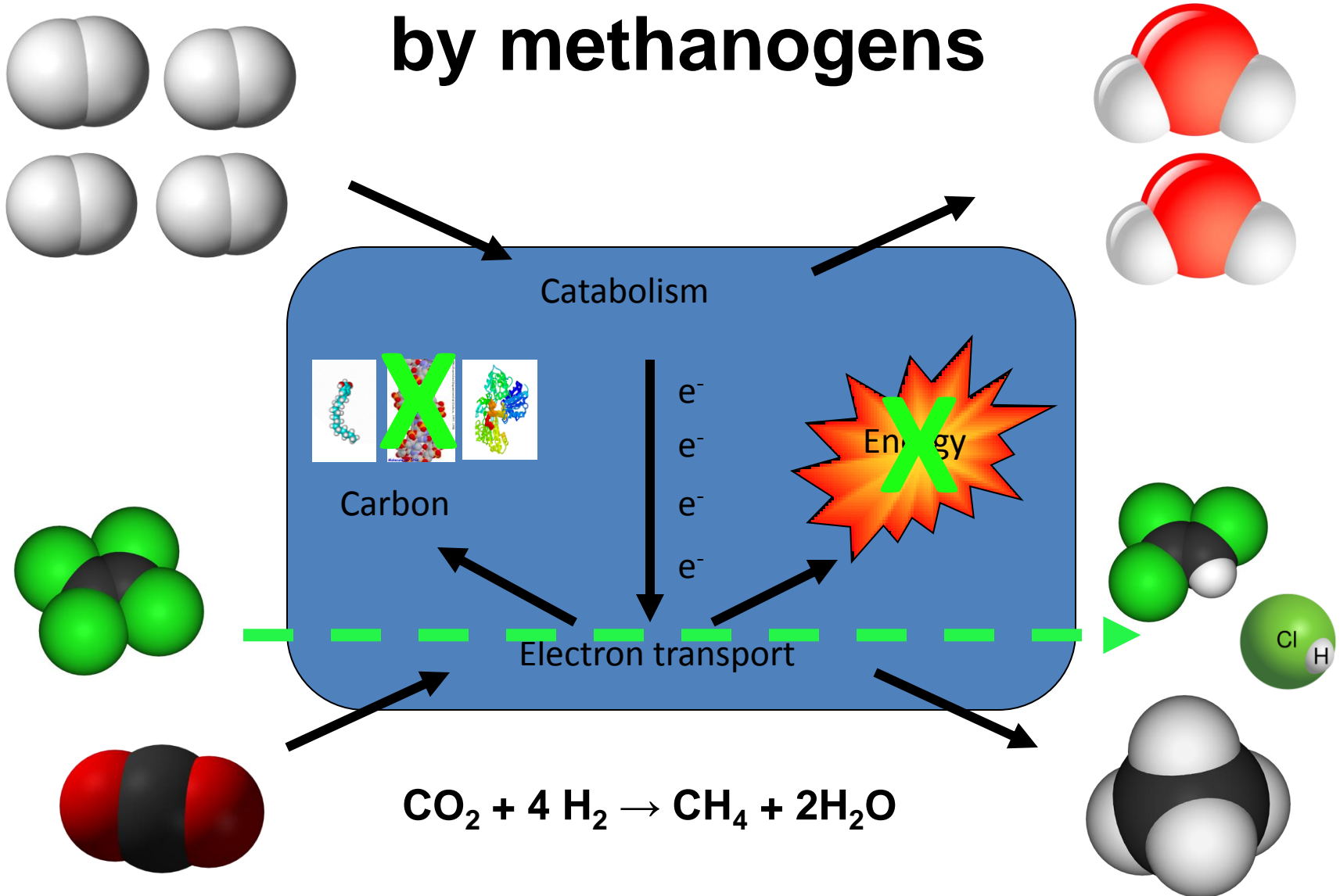
Aerobic TCE Cometabolism



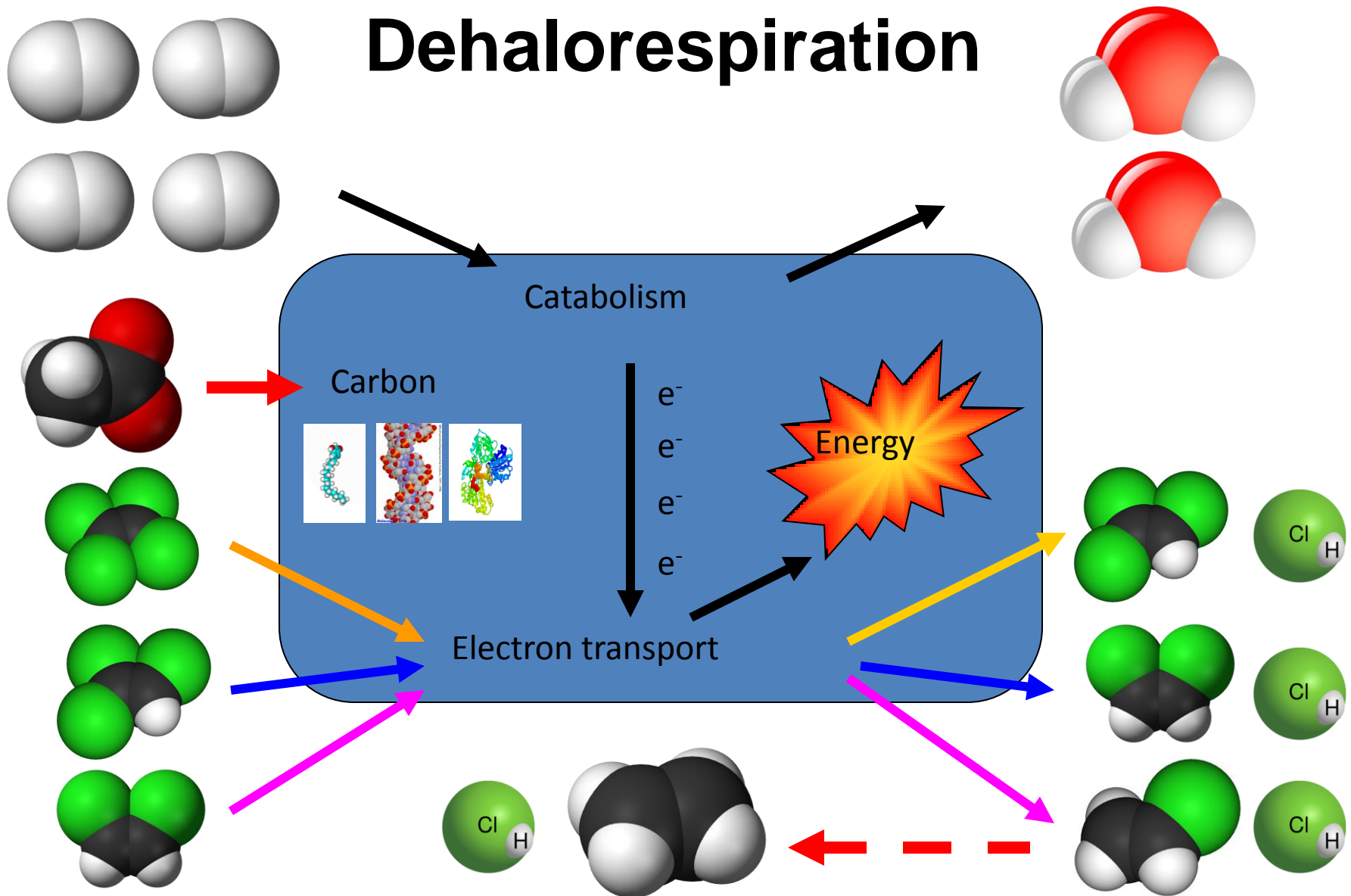
Methane-generating anaerobes (Methanogens)



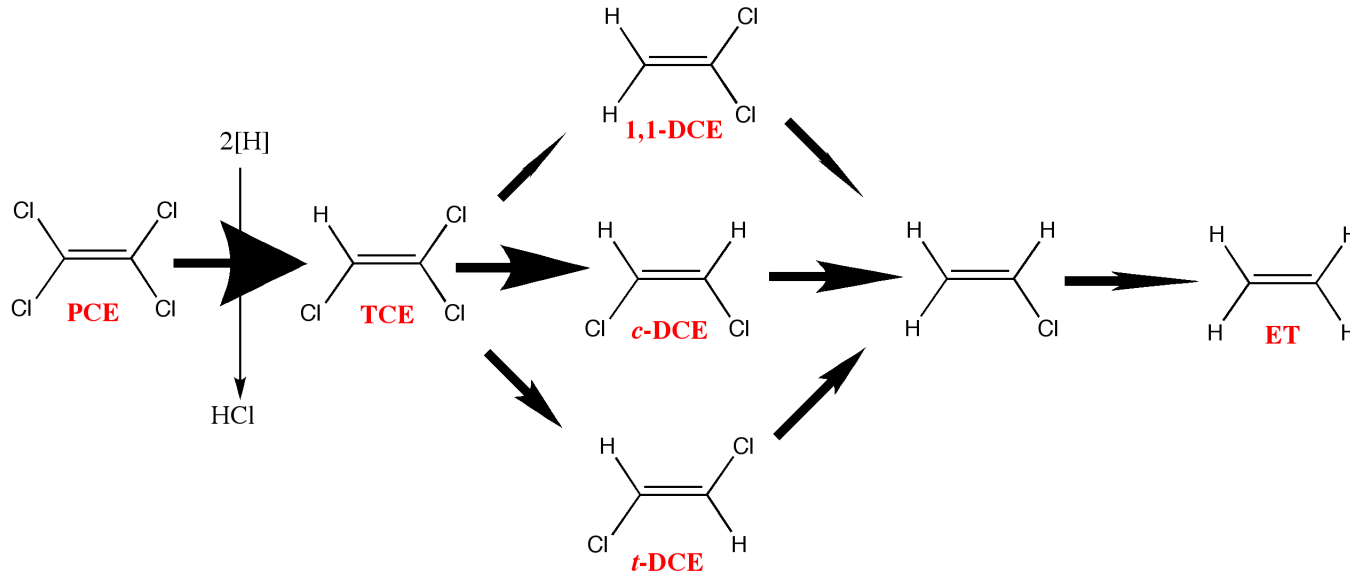
Cometabolic dehalogenation of TCE by methanogens



Reductive dehalogenation: Dehalorespiration



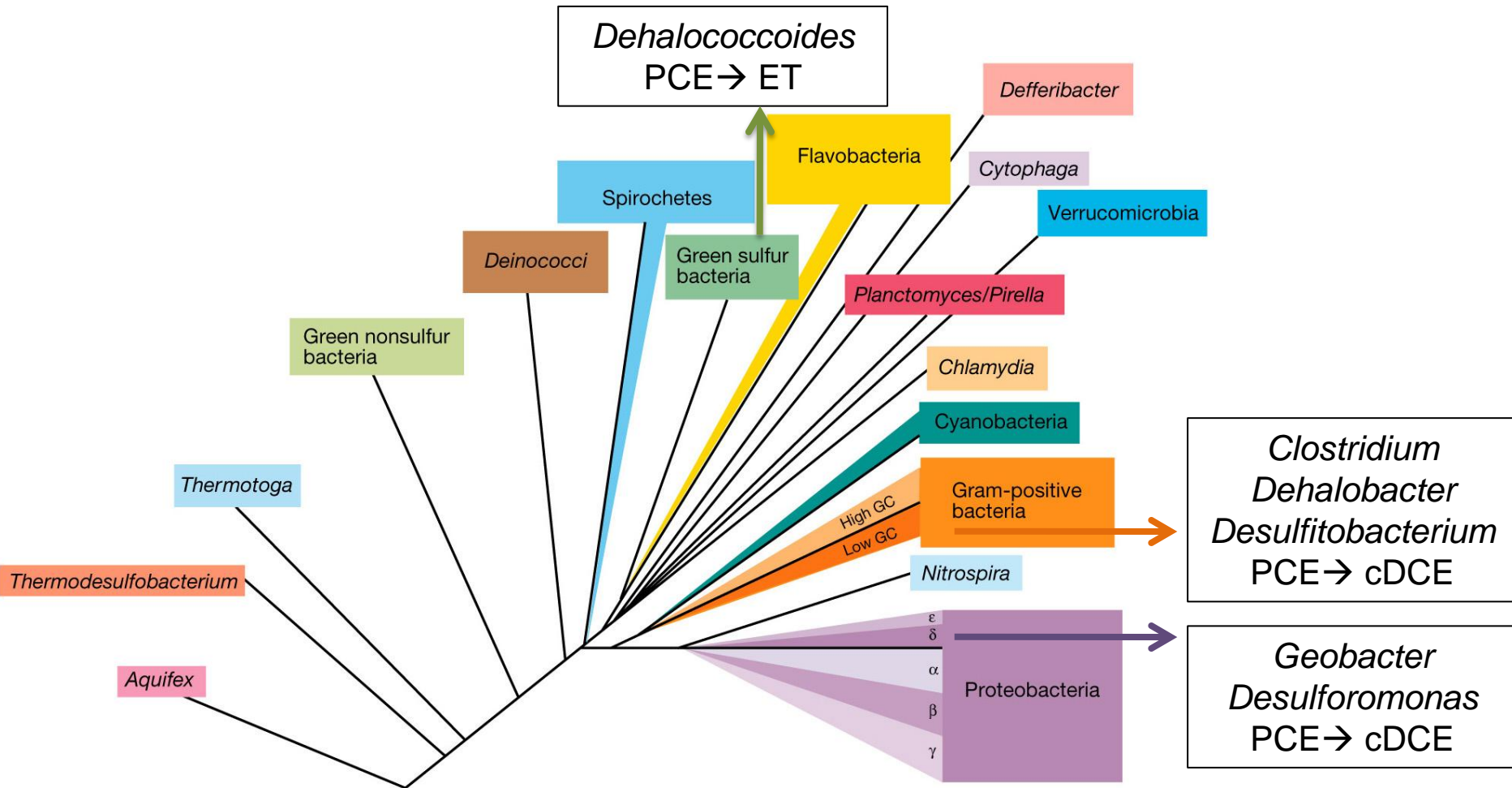
Pathway(s) of respiratory PCE reduction



- Respiratory process is much faster than anaerobic cometabolism
- Each reaction catalyzed by terminal reductase enzyme (e.g. *tceA*).
- Main flux is through *c*-DCE
- Rates typically decrease as acceptors become more oxidized.
- Process can often stall at *c*-DCE in the field.
- While ethylene is innocuous, VC is more toxic than PCE and TCE

| Organism | Strain | Phylogeny | PCE-TCE | TCE-cDCE | cDCE-VC | VC-ethene |
|---|--------------|----------------------------|---------|----------|---------|-----------|
| <i>Clostridium bifermentans</i> | DPH-1 | Low G+C G+ | Y | Y | N | N |
| <i>Dehalobacter restrictus</i> | PER-K23, TEA | Low G+C G+ | Y | Y | N | N |
| <i>Desulfitobacterium frappieri</i> | TCE1 | Low G+C G+ | Y | Y | N | N |
| <i>Desulfitobacterium metallireducens</i> | 853-15A(T) | Low G+C G+ | Y | Y | N | N |
| <i>Desulfitobacterium sp.</i> | Y51 | Low G+C G+ | Y | Y | N | N |
| <i>Desulfitobacterium sp.</i> | PCE1 | Low G+C G+ | Y | N | N | N |
| <i>Desulfitobacterium sp.</i> | PCE-S | Low G+C G+ | Y | Y | N | N |
| <i>Desulfitobacterium sp.</i> | KBC1 | Low G+C G+ | Y | N | N | N |
| <i>Enterobacter sp.</i> | MS1 | γ -proteobacteria | (Y) | (Y) | N | N |
| <i>Desulfomonile tiedjei</i> | DCB-1 | δ -proteobacteria | Y | Y | N | N |
| <i>Desulfuromonas michiganensis</i> | BB1 & BRS2 | δ -proteobacteria | Y | Y | N | N |
| <i>Desulfuromonas chlorethenica</i> | TT4B | δ -proteobacteria | Y | Y | N | N |
| <i>Geobacter lovleyi</i> | SZ | δ -proteobacteria | Y | Y | N | N |
| <i>Sulfurospirillum multivorans</i> | | ϵ -proteobacteria | Y | Y | N | N |
| <i>Sulfurospirillum halorespirans</i> | PCE-M2 | ϵ -proteobacteria | Y | Y | N | N |
| <i>Dehalococcoides ethenogenes</i> | 195 | Green non-sulfur | Y | Y | Y | (Y) |
| <i>Dehalococcoides sp.</i> | FL2 | Green non-sulfur | (Y) | Y | Y | (Y) |
| <i>Dehalococcoides sp.</i> | CBDB1 | Green non-sulfur | Y | N | N | N |
| <i>Dehalococcoides sp.</i> | GT | Green non-sulfur | N | Y | Y | Y |
| <i>Dehalococcoides sp.</i> | BAV1 | Green non-sulfur | N | N | Y | Y |
| <i>Dehalococcoides</i> (enrichment) | VS | Green non-sulfur | N | N | Y | Y |

Who are the PCE/TCE-respiring microorganisms?



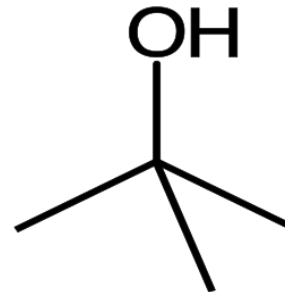
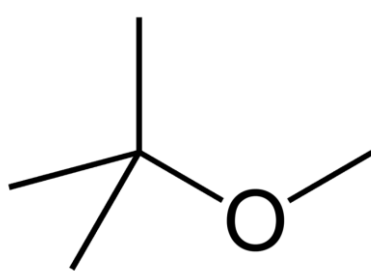
Fuel Oxygenates

A close-up photograph of a fuel nozzle, likely for a gas station. The nozzle has a green handle and a silver metal tip. The background is a blurred red color, possibly a gas station sign or wall. The text "Fuel Oxygenates" is overlaid at the top, and "MTBE and TBA" is overlaid at the bottom.

MTBE and TBA


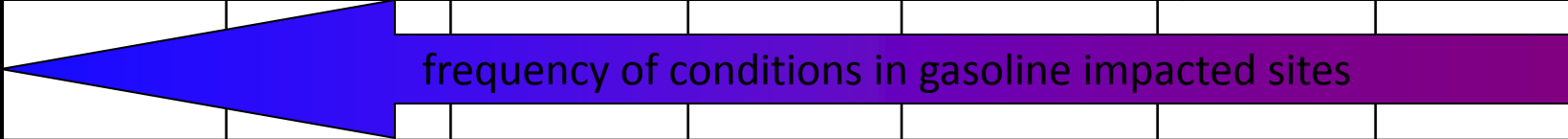
Overview of MTBE and TBA

- **MTBE: (methyl *tertiary* butyl ether)**
 - High volume chemical used in US as octane enhancer and fuel oxygenate from 1980s to mid 2000's. Limited current use in US. Widespread current use outside US.
- **TBA: (*tertiary* butyl alcohol)**
 - Limited early use as fuel oxygenate. Major biodegradation metabolite of MTBE

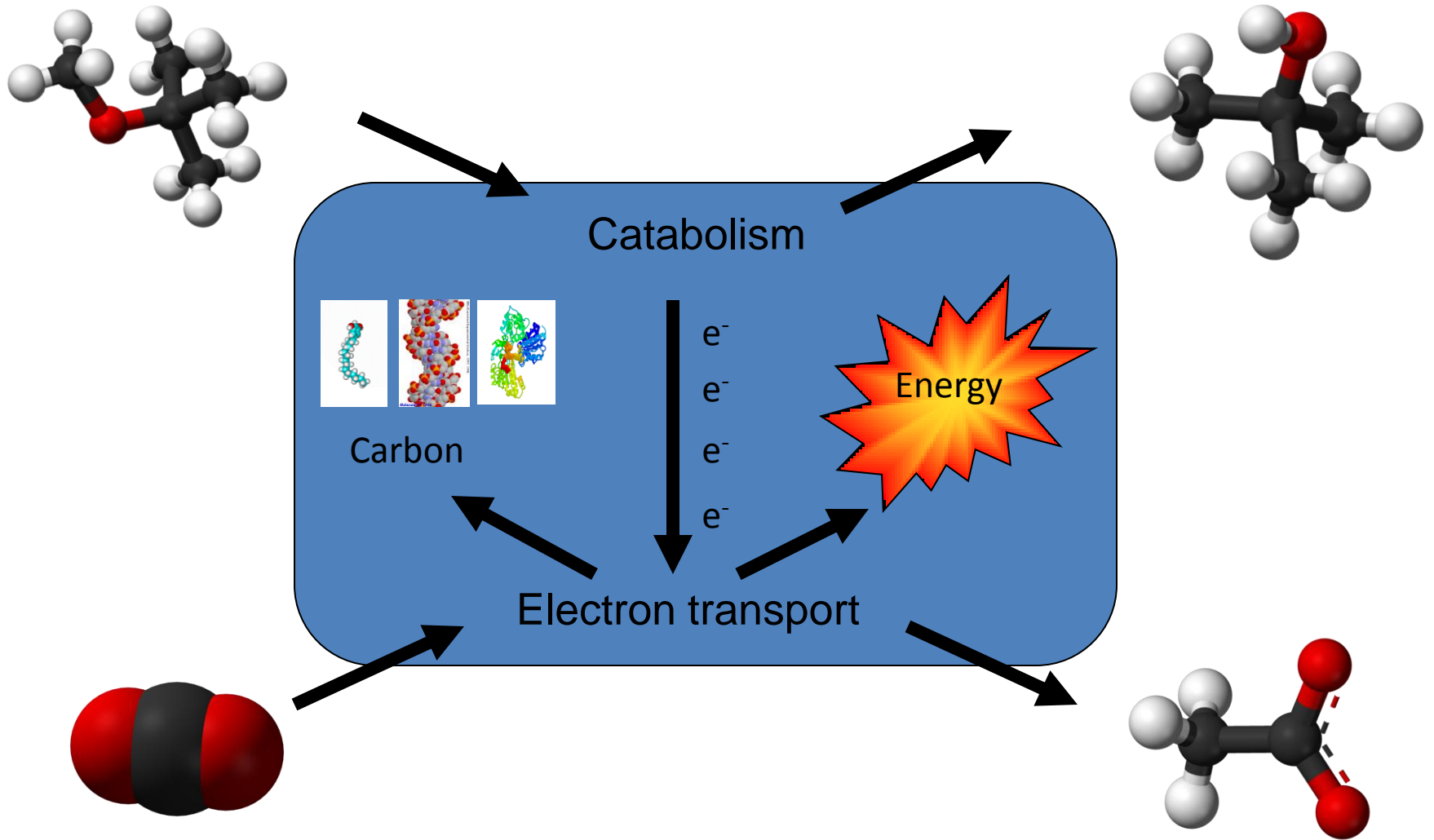


- Both common ground water pollutants, mainly from gasoline.
- Biodegrades under both aerobic and anaerobic conditions but no organisms known that utilize either compound as an electron acceptor.
- MTBE hydrolyzes to TBA at low pH. An analytical issue rather than environmental issue

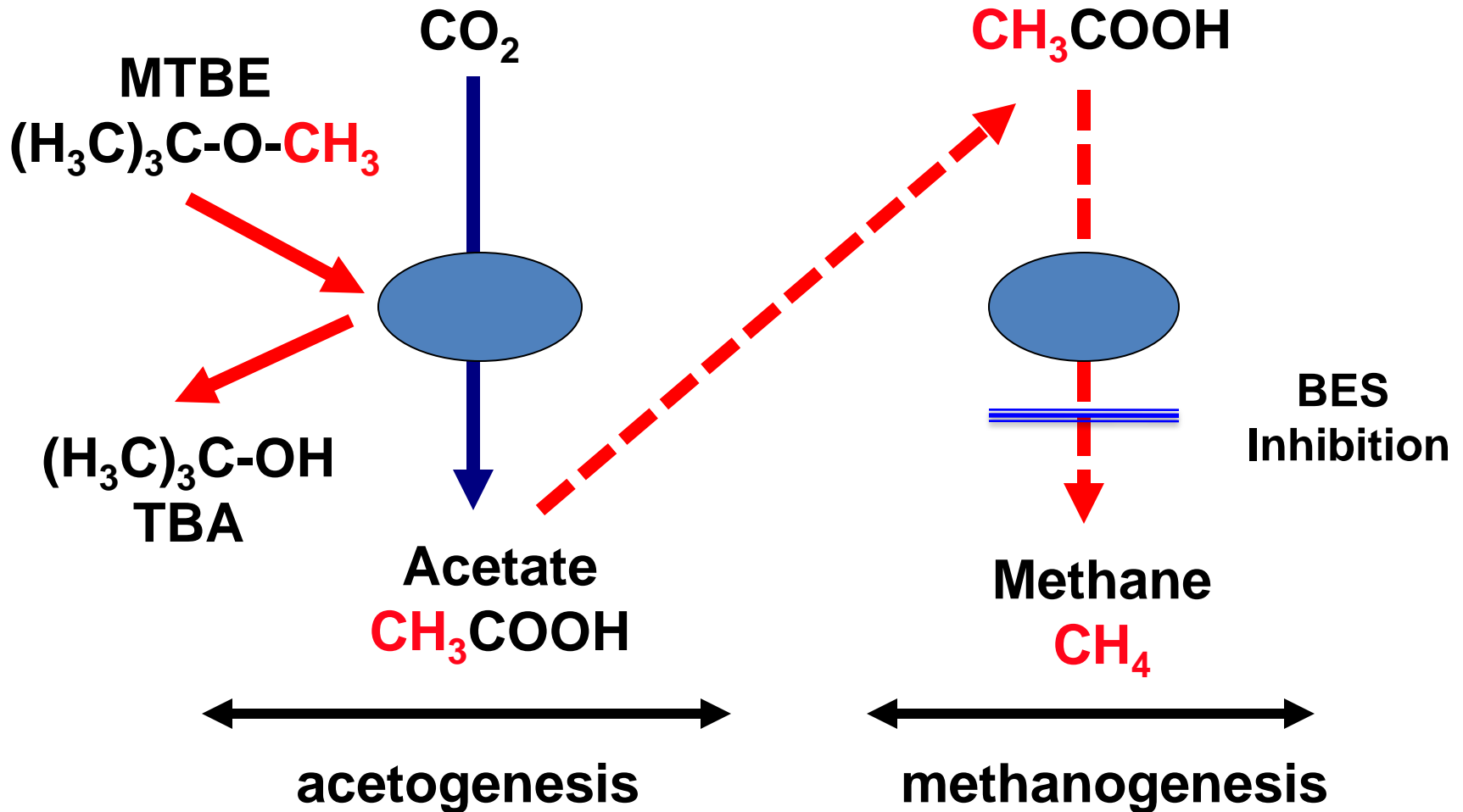
MTBE biodegradation summary

| Electron Acceptor | | | | | | |
|--|-----------------|-------------------------------|------------------|------------------|------------------------------|------------------|
| CO ₂ | CO ₂ | SO ₄ ²⁻ | Fe ³⁺ | Mn ⁴⁺ | NO ₃ ⁻ | O ₂ |
| CH ₄ | HAc | H ₂ S | Fe ²⁺ | Mn ²⁺ | N ₂ | H ₂ O |
|  degradation rate, energy yield, biomass generation | | | | | | |
|  frequency of conditions in gasoline impacted sites | | | | | | |
| MTBE | MTBE | MTBE | MTBE | MTBE | MTBE | MTBE |
| | | TBA | TBA | TBA | TBA | TBA |

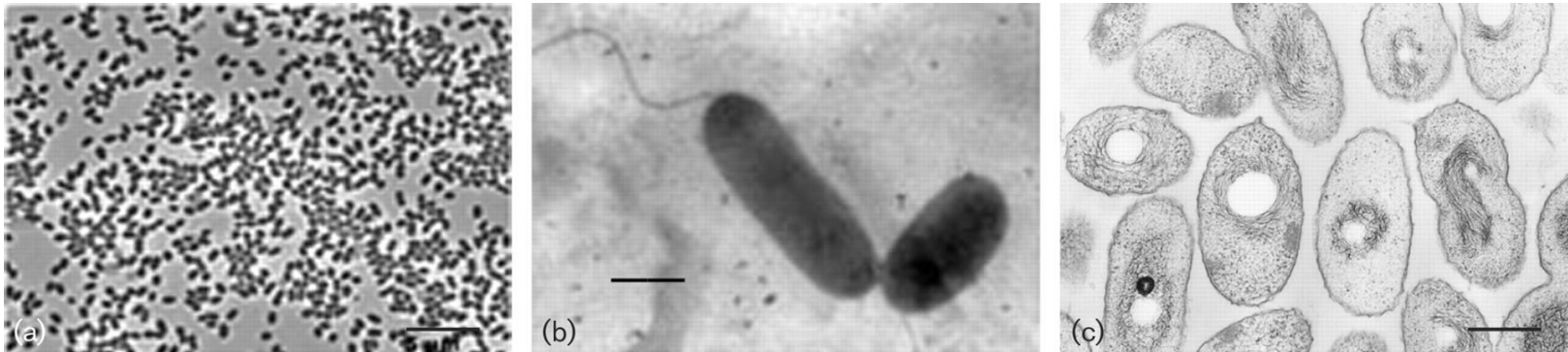
Acetate-generating anaerobes (Acetogens)



Acetogens, methanogens and anaerobic MTBE biodegradation?

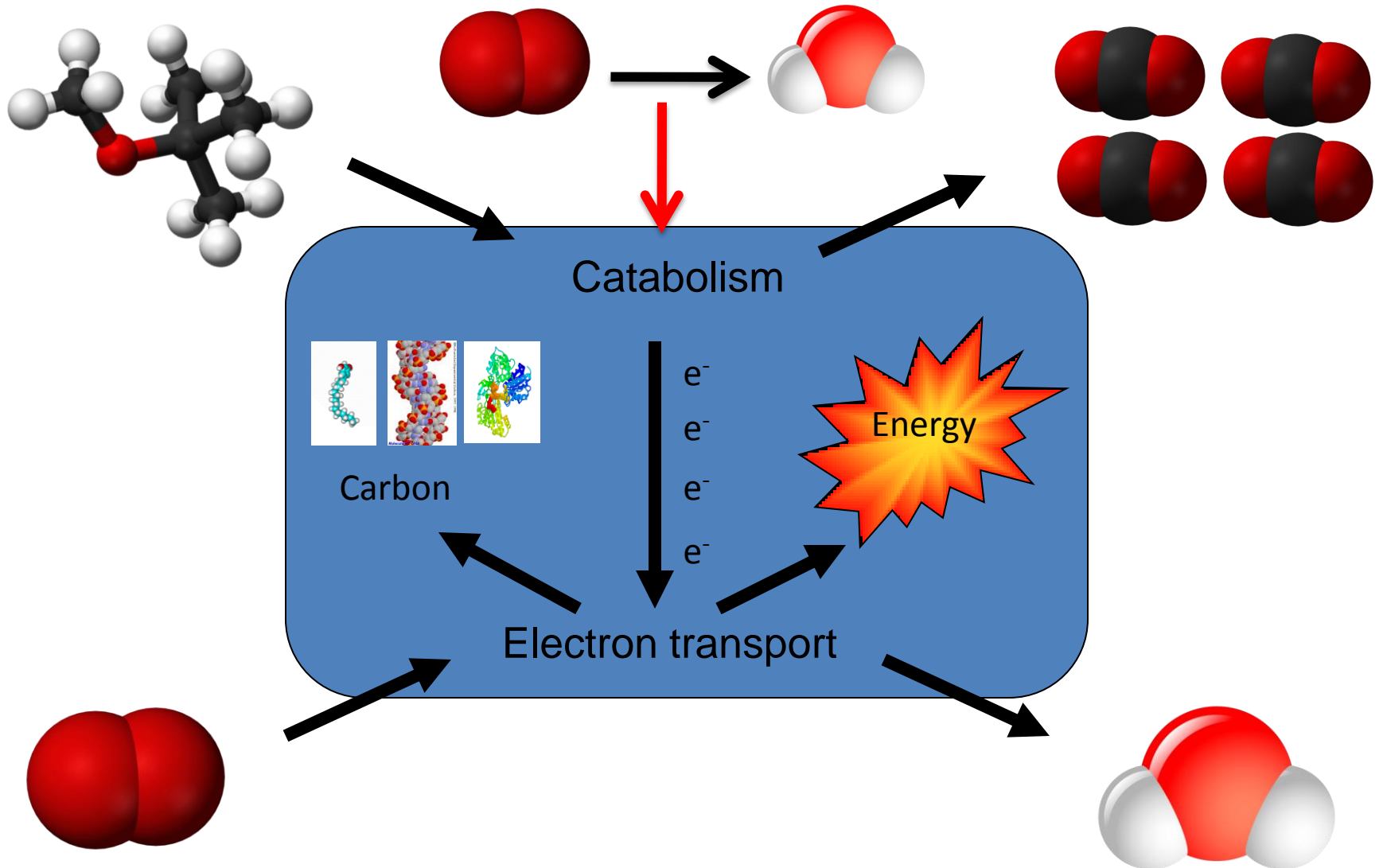


Aerobic MTBE/TBA metabolism

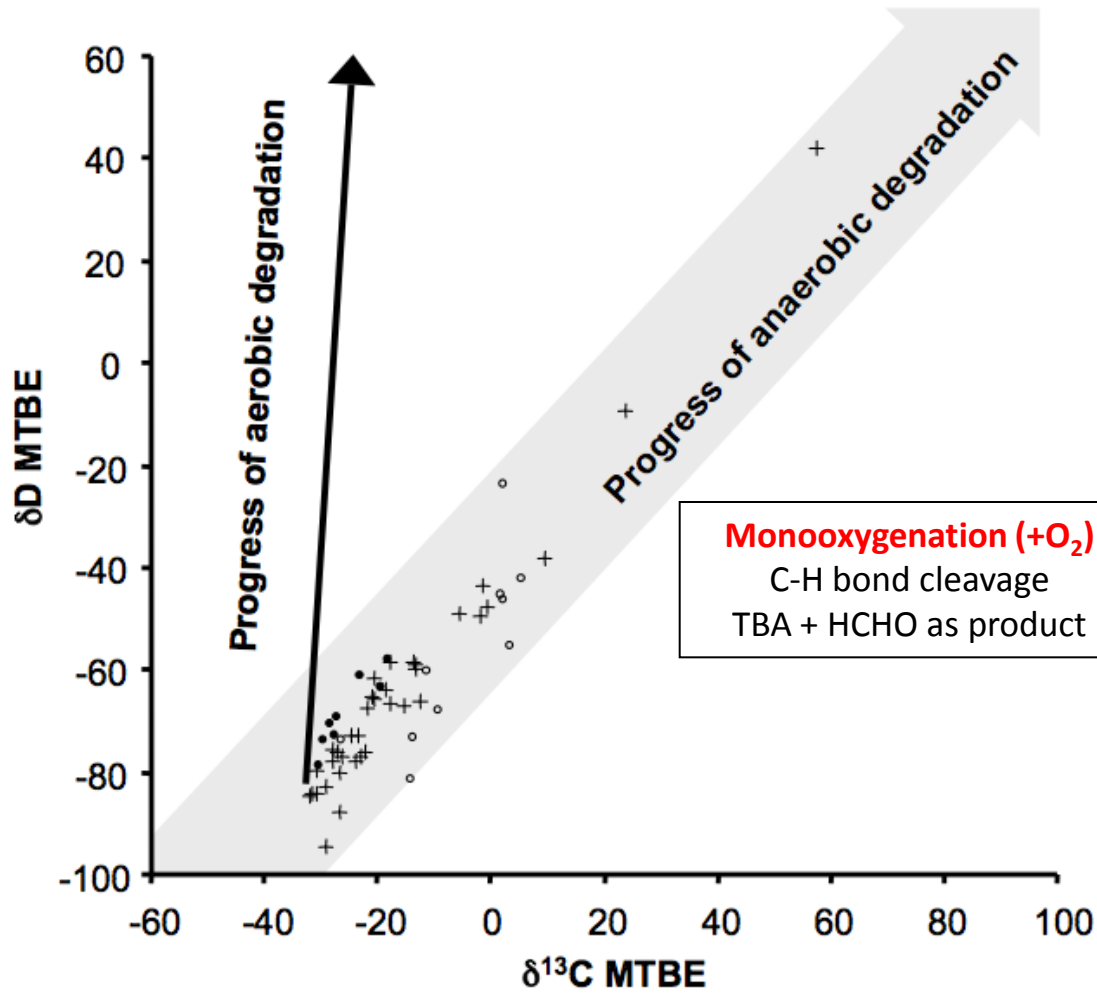


- ***Methylbium petroleiphilum* PM1:**
 - The DE 195 of the MTBE world!
 - Isolated from a peat moss biofilter.
 - Grows on MTBE, TBA, and several BTEX components.
 - Other MTBE-utilizing isolates are closely related (*Aquicola*, *Hydrogenophaga*)
 - Some closely-related strains lack the ability to oxidize MTBE but can grow on TBA.

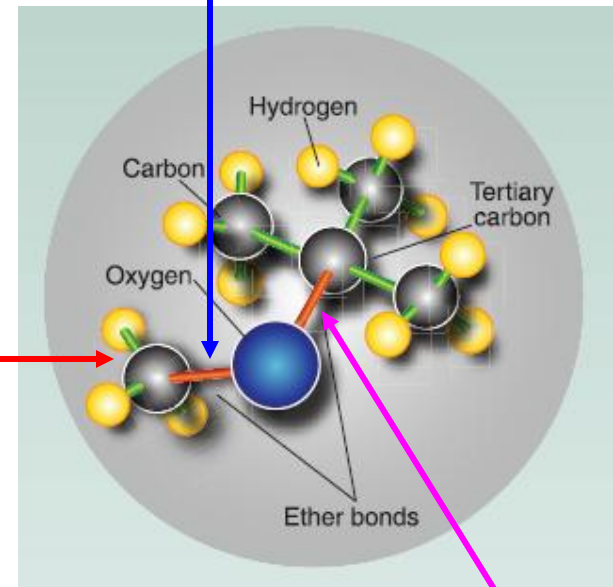
Methylibium petroleiphilum PM1



2D CSIA ($^{13}\text{C}/^{12}\text{C}$ & $^2\text{H}/^1\text{H}$) for MTBE



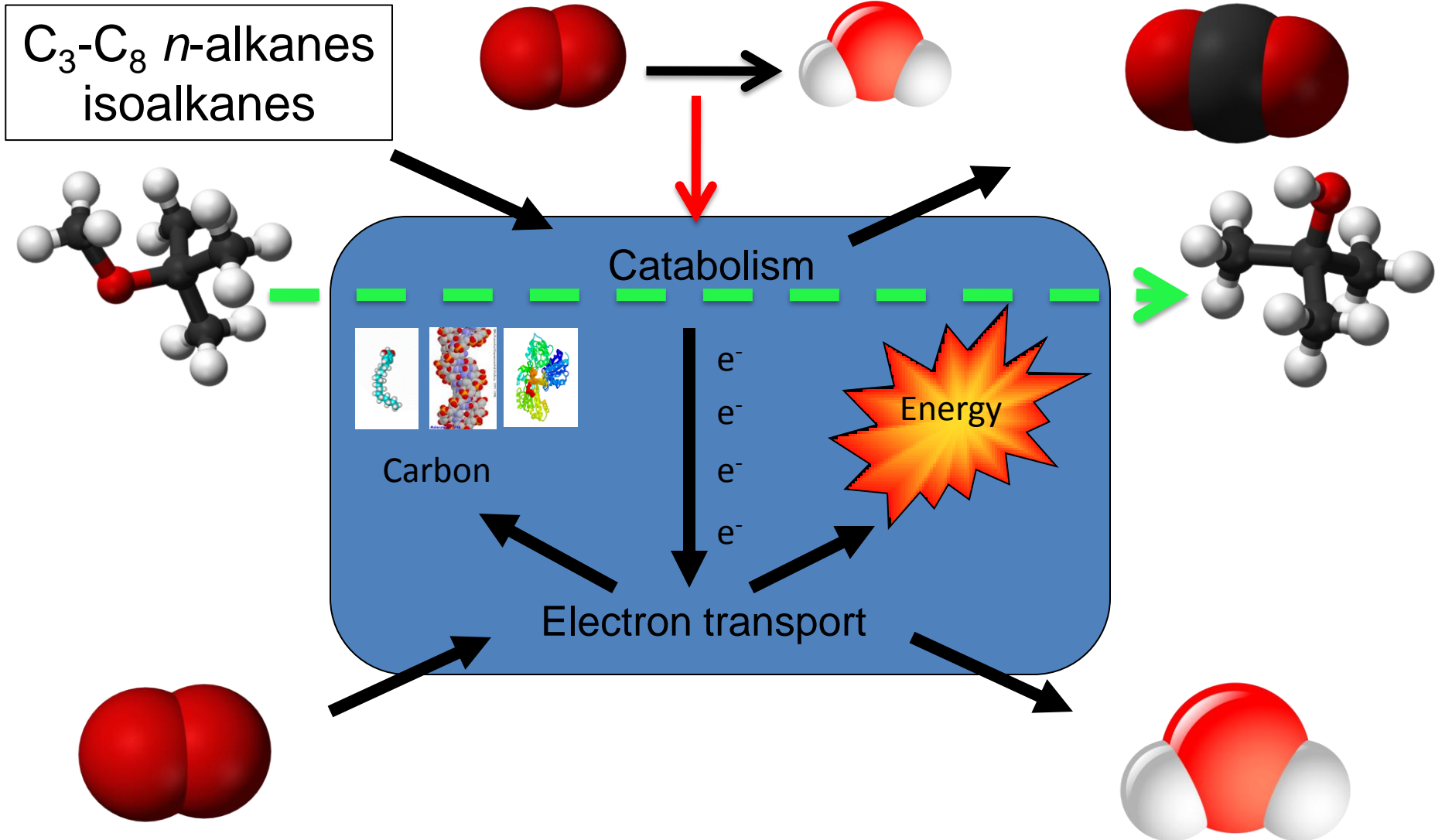
Demethylase ($-\text{O}_2$)
C-O bond cleavage
 $\text{X-CH}_3 + \text{TBA}$ as product



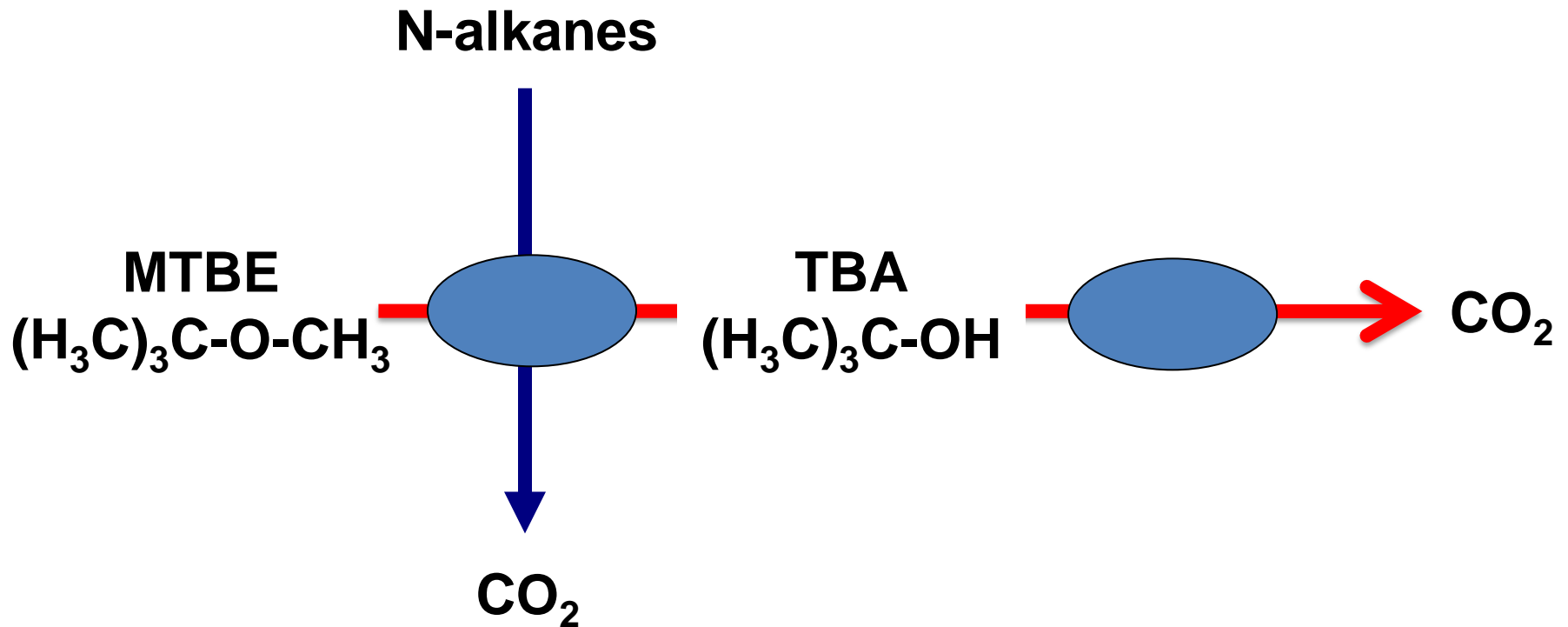
Monooxygenation ($+\text{O}_2$)
C-H bond cleavage
 $\text{TBA} + \text{HCHO}$ as product

Hydrolysis ($+/- \text{O}_2$)
C-O bond cleavage
 $\text{CH}_3\text{OH} + \text{TBA}$ as product

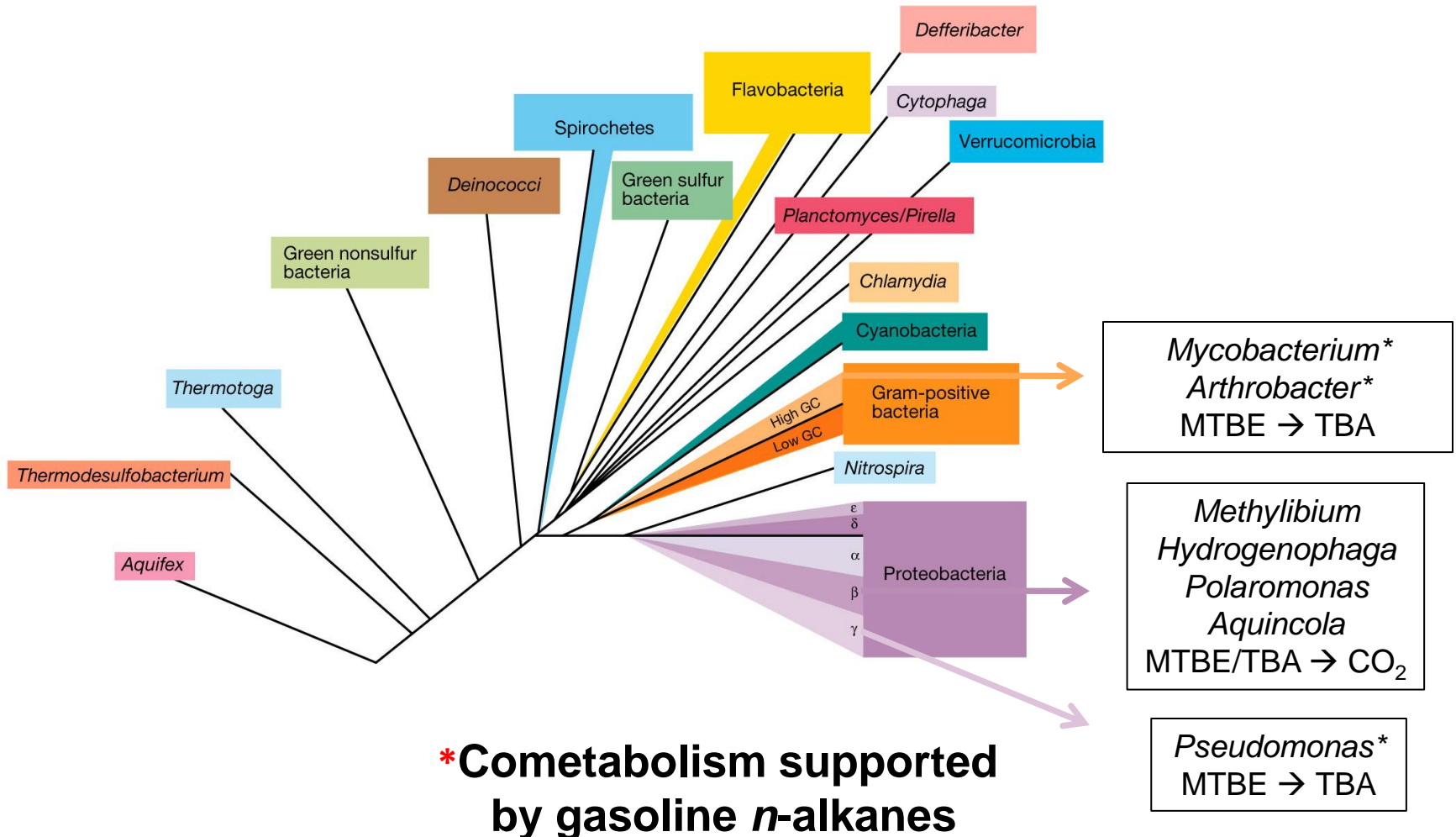
MTBE cometabolism by alkane-oxidizing bacteria



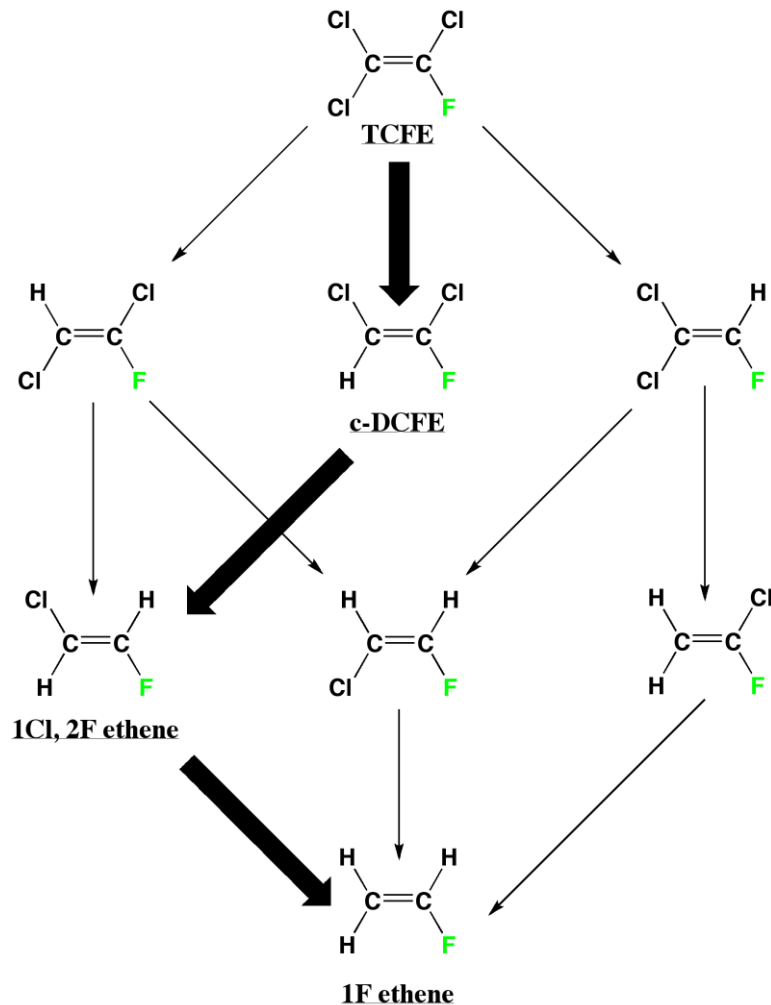
Commensalism and aerobic MTBE biodegradation?



Who are the aerobic MTBE-oxidizing microorganisms?



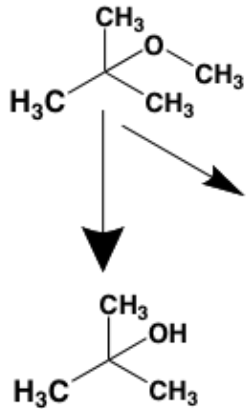
Fluorinated Enzyme Activity Probes



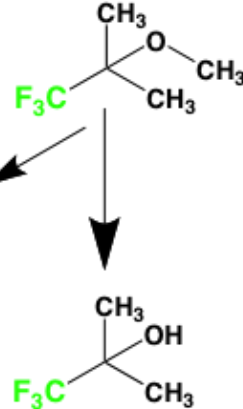
- Enzyme activity probes (EAPS) rely on lack of specificity of key enzymes in biodegradation pathways.
- Fluorinated analogs undergo same transformations as non-fluorinated compounds and at similar rates.
- Fluorine signature readily detectable, unique and retained during biodegradation.
- TCFE is analog for PCE/TCE
- VF also investigated for aerobic and anaerobic VC degradation.
- *In situ* applications using Push-Pull tests

Fluorinated MTBE analogs

Methyl *tertiary* butyl ether
MTBE

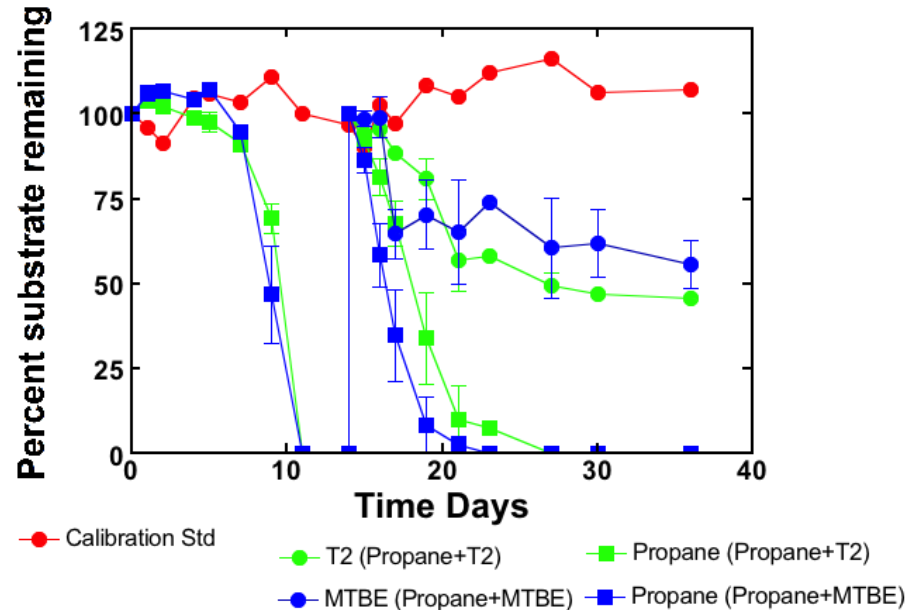


2-trifluoromethyl-2-methoxypropane
T2



tertiary butyl alcohol
TBA

2-trifluoromethyl-2-propanol
TFMP



- T2 serves as an accurate analog of MTBE in cometabolic microcosms.
- TFMP used as a probe for pathway of MTBE/TBA oxidation.
- Both are useful as reactive probes for *in situ* determination of potential and rates of MTBE/TBA degradation.